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**SECTION 15.1 WATER AND ITS PROPERTIES**  
(pages 445–449) This section describes the properties of water in the liquid and solid states and explains how hydrogen bonding affects the surface tension and vapor pressure of water. **Water in the Liquid State** (pages 445–447)

- 1.

### **SECTION 15.1 WATER AND ITS**

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**PROPERTIES (pages  
445-449)**

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## Aqueous Systems Chapter 15 “Water and Aqueous Systems”

The Water Molecule: a Review. • Water is a simple tri-atomic molecule, H<sub>2</sub>O.

- Each O-H bond is highly polar, because of the high electronegativity of the oxygen (N, O, F, and Cl have high values)
- bond angle of water = 105°.

## **Chapter 15 Water and Aqueous**

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Chemistry, Chapter 15,  
Water and Aqueous  
Systems. surface  
tension. surfactant.  
aqueous solution.  
solvent. the inward  
force or pull that tends  
to minimize the surface  
ar.... any substance  
that interferes with  
hydrogen bonding  
between wa.... is water  
that contains dissolved  
substances.

**chemistry chapter**

*Page 8/26*



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these activities to yourself study the vocabulary and major concepts presented in this chapter.

heterogeneous mixture containing particles that are small enough to remain dispersed in the solvent and do not separate on standing.

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17 in your books“Water  
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2. Section 15.1 Water  
and its

PropertiesOBJECTIVES:

- Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding. •

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15.3 Heterogeneous

Aqueous Systems -

15.3 Lesson Check -

Page 507: 21 Answer

Colloids cannot be

separated through

filtering because their

particles are smaller

than those of

suspensions and they

do not settle out over

time.

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resource, we all  
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3 Heterogeneous  
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Aqueous Systems 1  
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Systems. Pre-AP  
Chemistry ; Charles  
Page High School ;  
Stephen L. Cotton; 2  
Chapter 15 Review.  
Surface tension is the

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Water. How does the surface tension of water compare with the surface tensions of most other liquids? Which type of mixture(s) exhibit the ...

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3 Heterogeneous  
account; E-mail to a  
friend; Find other  
activities ... a relatively  
strong intermolecular  
force responsible for  
water's high surface  
tension: suspension: a  
mixture from which  
particles settle out  
upon standing:  
aqueous solution: a  
solution in which the  
solvent is water:  
solvent: the ...

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Chapter 15 - Water and  
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15.1 Water and Its  
Properties - 15.1

Lesson Check - Page

493: 2 Answer The

structure of ice is a  
regular open

framework of water  
molecules arranged

like a honeycomb.

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SECTION 15.1 WATER  
AND ITS PROPERTIES  
(pages 445–449)

CHAPTER 15, Water and  
Aqueous

Systems(continued) 6  
Circle the letter next to  
each sentence that  
describes a result of

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## Aqueous Systems 3 Heterogeneous

the surface tension of water a In a full glass of water, the water surface seems to bulge over the rim of the glass b Water beads up into

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