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Background: This text was used in Spring 2014 for Donald P. Visco's undergraduate course in Equilibrium Thermodynamics taught within the Chemical Engineering Department at The University of Akron. This course was taken

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by mostly STEM and Honors students. This was the first semester of the text's publication. The class completed Chapters 1-14

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- Chemical equilibrium
- no tendency for a
species to change
phases or chemical
react •

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equilibrium - a system
that is in mechanical,
thermal, and chemical
equilibrium • Phase
equilibrium - a system

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with more than one
phase present that is in
thermal and
mechanical

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Part I clearly introduces the laws of thermodynamics with applications to pure fluids. Part II extends thermodynamics to mixtures, emphasizing phase and chemical equilibrium.

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Throughout, Matsoukas focuses on topics that link tightly to other key areas of undergraduate chemical engineering, including separations, reactions, and capstone design.

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The approach taken stresses problem-solving, and draws from "best practice" engineering teaching strategies. FUNDAMENTALS OF CHEMICAL ENGINEERING THERMODYNAMICS uses examples to frame the importance of the material. Each topic begins with a motivational example that is investigated in context to that topic.

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