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$\Delta H$  (H)  $-1 = -840$  kJ mol<sup>-1</sup> A.  $-190$  B.  $-95$  C.  $+6$  D.  $+95$   $-1$  7 2 5 7  $-1$  5 Which is a correct definition of lattice enthalpy? A. It is the enthalpy change that occurs when an electron is removed from 1 mol of gaseous atoms. B. It is the enthalpy change that occurs when 1 mol of a compound is formed from its elements.

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