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Isotope and Ions Practice Worksheet Name - Yumpu

Answer: 126.86 amu 10. The natural abundance for boron isotopes is 19.9% ^{10}B and 80.1% ^{11}B . Calculate boron's atomic mass. Answer: 126.86 amu 11. Hydrogen is 99% ^1H , 0.8% ^2H , and 0.2% ^3H . Calculate its average atomic mass. Answer: 1.21 amu 12. Rubidium is a soft, silvery-white metal that has two common isotopes, ^{85}Rb and ^{87}Rb . If the ...

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